

units. On the other hand, in the case of triscarbazoles linked via C-N bonds, the combination of inductive effects and molecular orbital interactions tunes the frontier level energies and, interestingly, gives rise to an ambipolar character. In the C-N linked systems, the lowest triplet states are characterized mainly by an electronic transition localized within the central carbazole, while in the C-C linked compounds it is the longest oligo-*para*-phenyl segment to be found in the chemical structure that defines the lowest triplet transition. When the N-H group of the central carbazole unit is replaced by other groups [O, S, CH₂, C(CH₃)₂, C(CH₃) (CF₃), and C(CF₃)₂], the HOMO/LUMO energies fluctuate substantially in the absence of the side carbazoles, but these variations are significantly reduced in their presence; also, the singlet-triplet energy differences decrease substantially when going from the isolated central unit to the triscarbazole-like derivatives.

Keywords: blue OLEDs; ambipolar hosts; triscarbazoles; singlet-triplet energy differences; density functional theory (DFT) calculations



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